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Dutch Institute for
Fundamental Energy Research

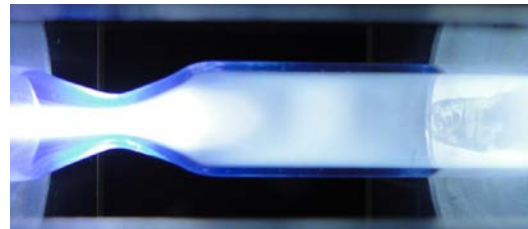


University of Stuttgart
Germany

CO₂-Neutral Fuels

Adelbert Goede

Waldo Bongers, Martijn Graswinckel, Erik Langereis and Richard van de Sanden



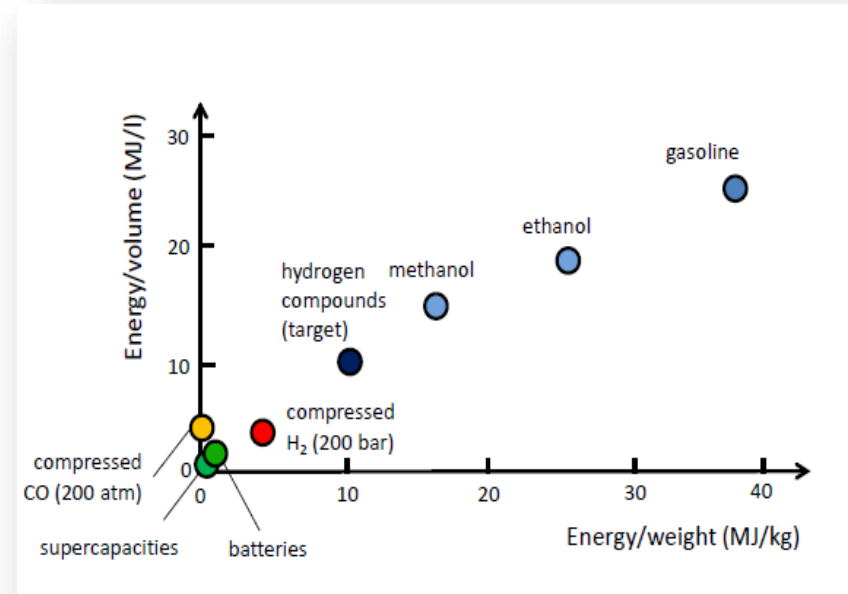
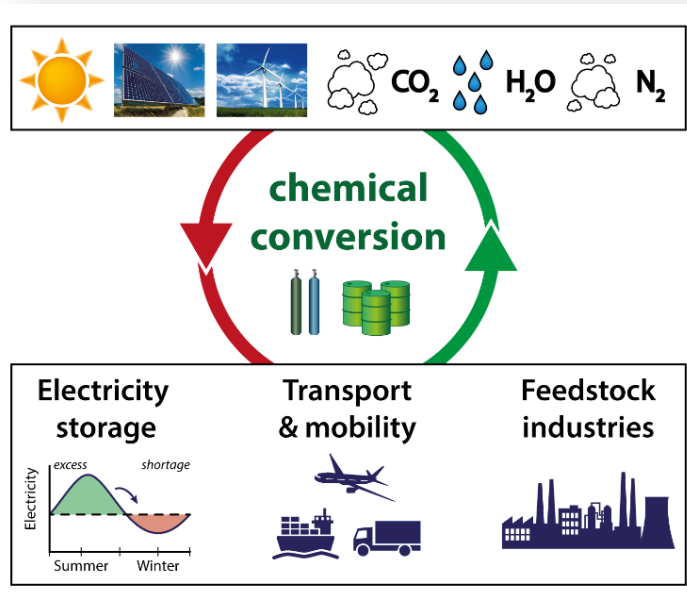


CO₂ Neutral fuels: What are they?

Hydrocarbons synthesised from water and air

- powered by Renewable Electricity
- CO₂ recirculated after use

Characterised by high energy density and existing infrastructure





Why CO₂ Neutral Fuels?

- UNFCCC Paris 2015: CO₂ emission curbed by 80%-95% in 2050 vs. 1990
- Transport target EU-RED 2050:
 - 60% CO₂ emission reduction
 - Aviation 40% RE by 2050 (~2% of global CO₂ emission)
- Transport makes up ~25% EU primary energy consumption

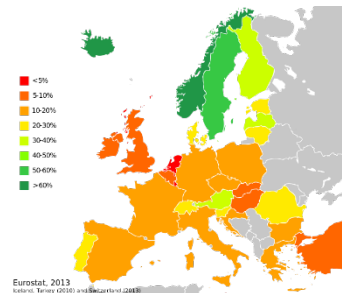
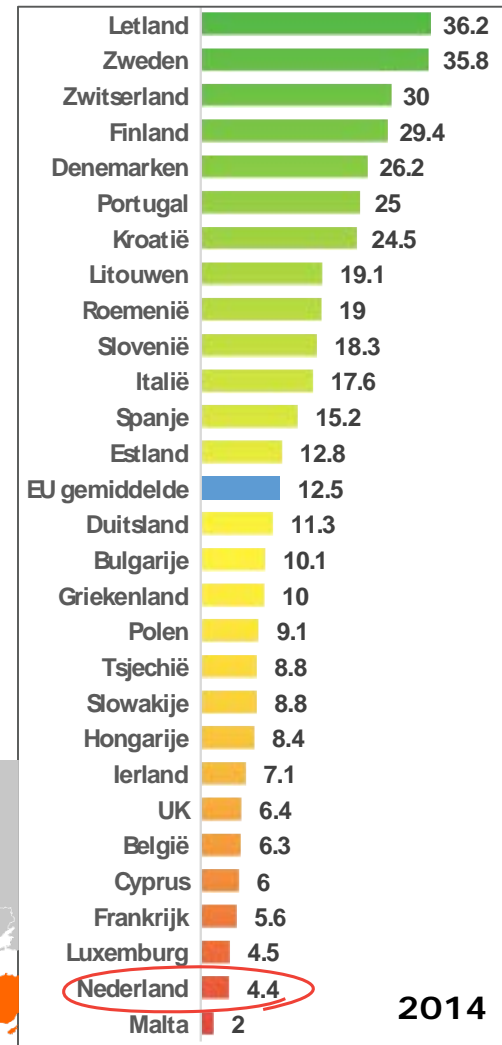
→ Batteries and Hydrogen emission free, but

→ low volumetric energy density
(hence mobility range limit)

→ Bio Fuels

→ limited by Fuel vs. Food/Flora trilemma
(jet fuel alone 5M barrels/day kerosene)

% renewables EU



2014



Aimed at phasing out Fossil and Nuclear and replace by Renewables

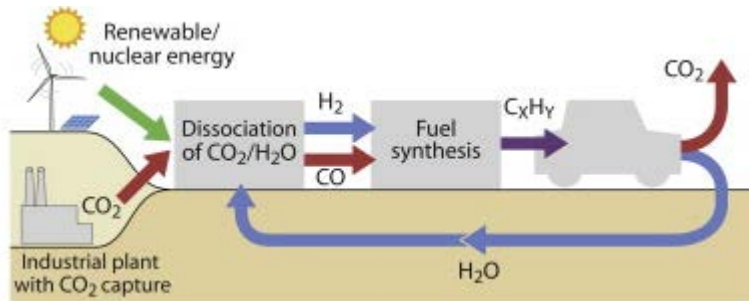
- Goal 80% electricity from Renewables in 2050
- Presently: 30% RE, 45GW installed wind power
- 30% RE on average: sunny and windy day 90% RE (8 May 2016). Cloudy and Calm: fossil coal, lignite and gas scrambled to cover demand
- Result: Dynamic grid control problem and CO₂ emissions up
- Objectives: clean, affordable, reliable energy supply not met
- Subsidised through guaranteed 20 yr electricity price (20B€/yr)
- Reform now drafted: Auction on generating capacity
- Needed: System Approach of generation, transmission, distribution to customers consistently managed
- Includes Energy Storage: BMBF Kopernikus Programme P2X



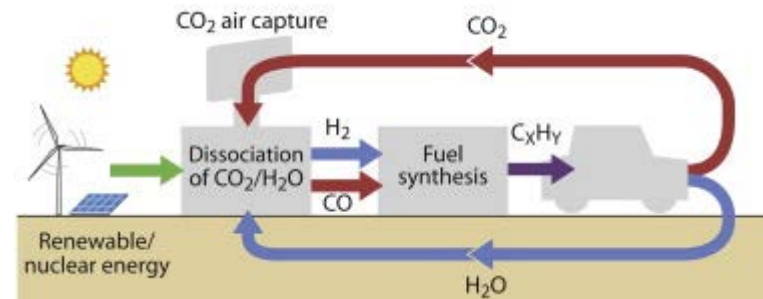
Carbon neutral fuel cycle: P2X – CCU

Graves et al., Ren. Sustain. Energy Rev. **15**, 1, (2011)

Point source capture of fossil CO₂
→ not climate neutral, emission delayed



Direct air capture of CO₂
→ climate neutral fuel cycle



Power-to-X

X = gas or liquid fuel or chemicals

P2X + CCU

CCU: carbon capture and utilisation

P2X is most critical part both technically and economically

Technology benchmark: costs of H₂

- Electrolysis >6 €/kg H₂ (fossil fuel <1 €/kg H₂)
- CO₂ capture: point source 40 €/tonne, direct air 400 €/tonne

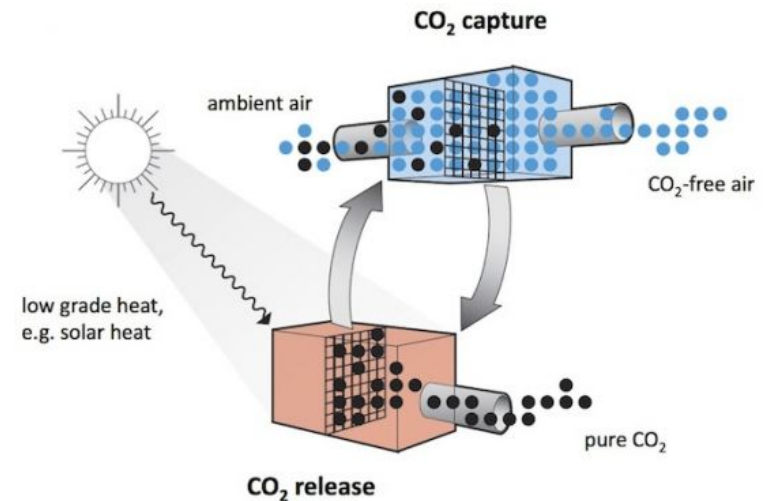


Direct air capture of CO₂

Klaus Lackner Tree captures 1 tonnes CO₂/day. Wet-dry cycle epoxy. To supply wind farm with CO₂ to be split by P2X, an approximately equally sized Lackner wood is needed.



Klaus Lackner
Center for negative carbon emissions



First commercial CO₂ air capturing plant, Hinwil (CH), capacity 900 ton CO₂ /yr, swept area 6x12m² amine granules



Coupling Electricity, Gas and Oil infrastructure

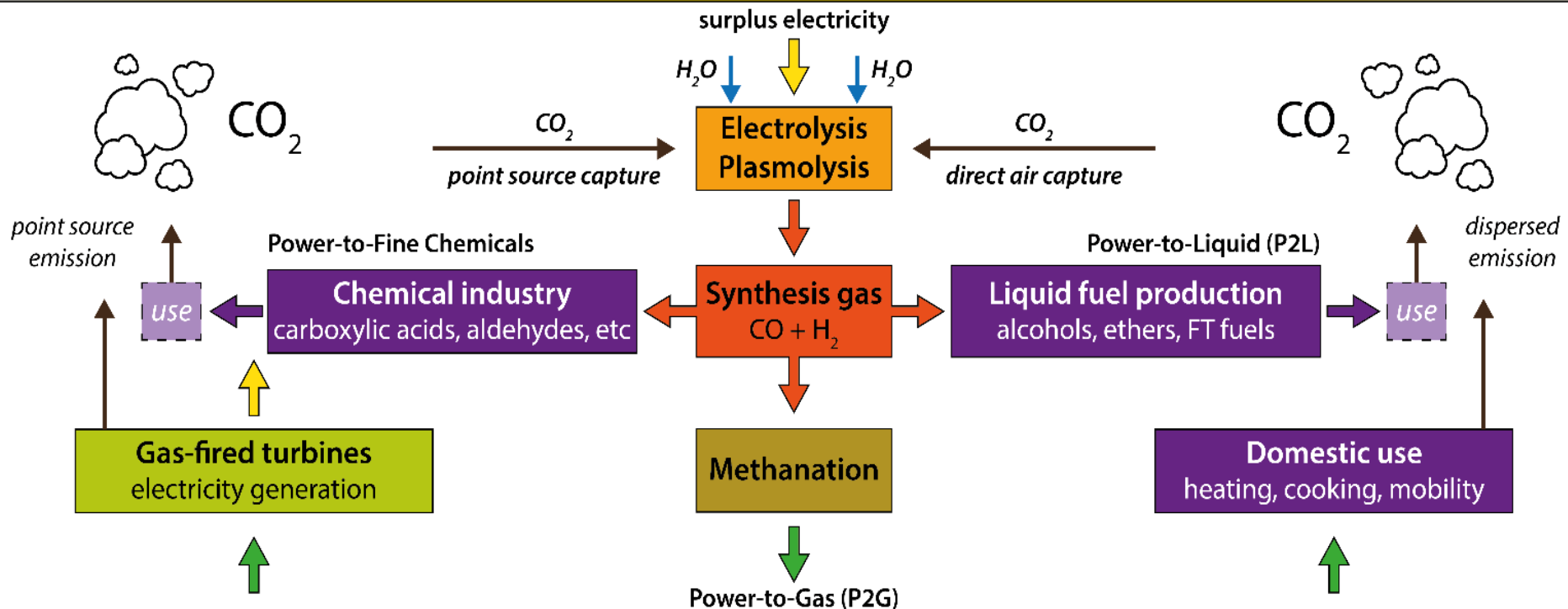
Surplus renewable electricity expected 2050:

Germany 110-148 TWh, France 44-91 TWh, The Netherlands 30-55 TWh

Energy Storage capacity NL gas net 552TWh

Renewable energy: intermittency, mismatch supply and demand

Electricity Grid - sustainable electricity



Gas Grid - synthetic natural gas (SNG)

Energy storage, transport, distribution



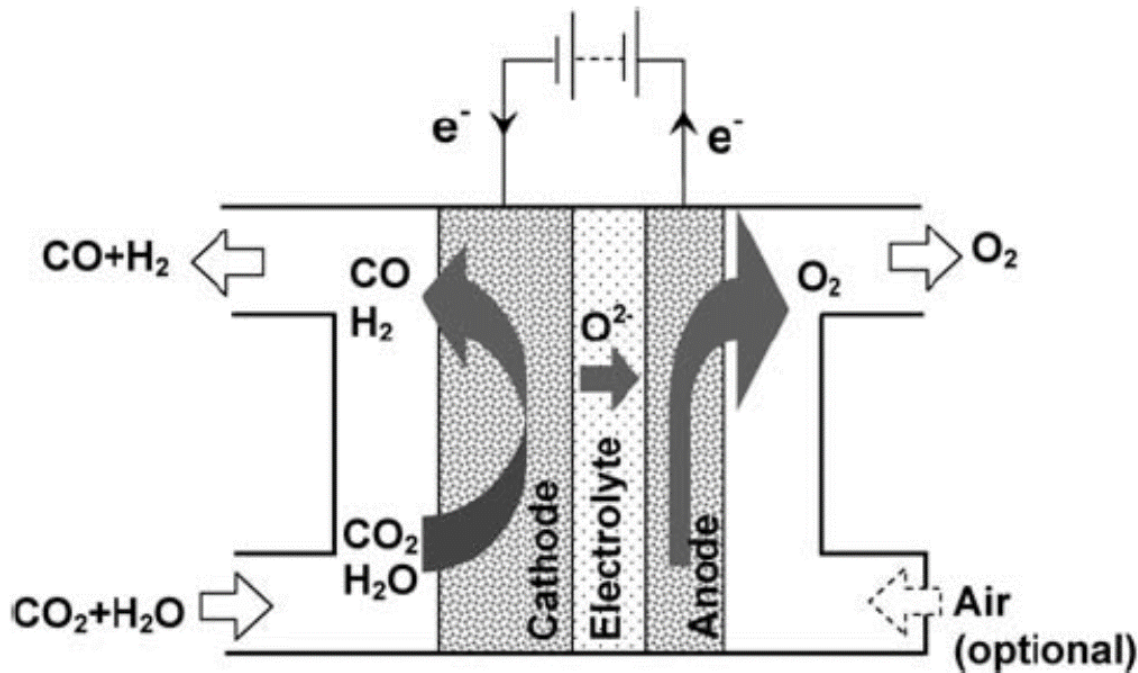
Splitting H₂O and/or CO₂ by electrolysis

- **Alkaline** electrolyte (100 yrs large scale, mature technology)
 - Power density low ($< 0.5\text{W}/\text{cm}^2$)
 - Hydrogen output pressure low ($< 30\text{bar}$)
 - Safety (caustic electrolyte)
- **PEM** (polymer electrolyte membrane), pre-commercial
 - Power density $\sim 1\text{W}/\text{cm}^2$
 - Rapid dynamic response
 - Degradation membrane
 - Catalyst material Pt, Ir (Scarce)
 - MW unit (Siemens)
- **SOEC** (solid-oxide electrolyser cell)
 - High power density, energy efficiency, output pressure
 - High Temperature operation (800°C , pressure 50-100 bar)
 - Co-electrolysis H₂O and CO₂
 - Degradation under high current density operation
 - Upscaling from kW range hampered



SOEC Co-Electrolysis at DTU

- External DC voltage pumps O^{2-} ions from porous **cathode** (Ni/YSZ)
- through dense solid **electrolyte** (YSZ = Yttrium Stabilised Zirconia)
- to porous **anode** ($La_{1-x}Sr_xMnO_3/YSZ$) at high temperature ($800\text{ }^\circ\text{C}$)

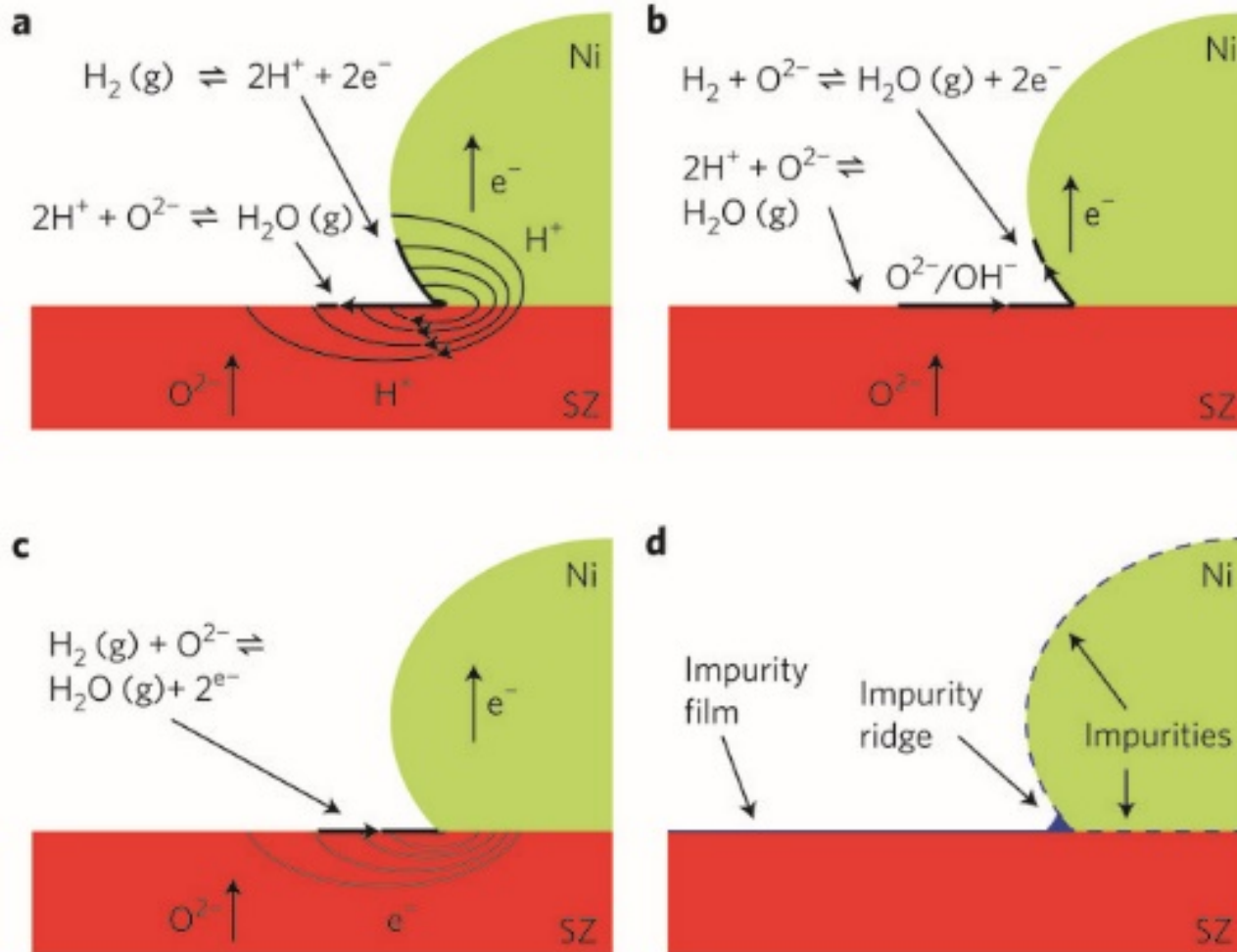


DTU Technical
University of
Denmark
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Electrolysis is a surface process happening at the nm interface between electrode, electrolyte and catalyst a 3-phase boundary phenomenon not understood



John Irvine et. al. Nature Energy 1, p.1-23 (2016)



Fuel Cell mode

- H_2 absorption and dissociation on Ni surface to form H^+ migrating to site for water formation
- Migration of O^{2-} or OH^- along electrolyte and Ni surface to form water
- Water formation at electrolyte surface and electron transport through the electrolyte
- Role of impurities (glassy Silicate) at the 3PB to block the water formation reaction



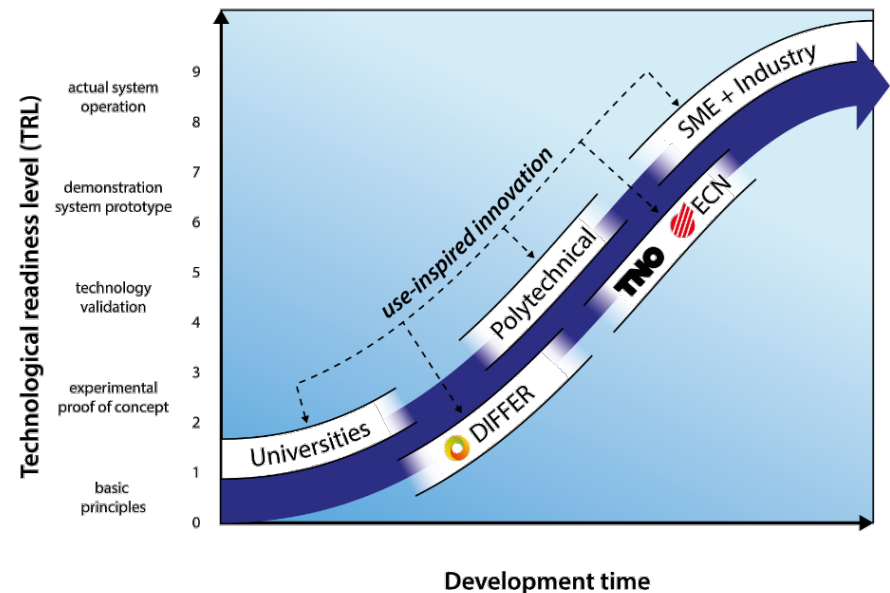
SCIENCE FOR FUTURE ENERGY

Mission: Basic scientific research into
Fusion Energy and **Solar Fuels**,

Based on in house **high-quality technical infrastructure**,
collaboration with Academia, National Research Organisations and Industry,
building a **national community** in energy research.



*Relocated mid 2015
University Campus Eindhoven*



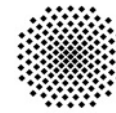


Why plasma for CO₂ conversion?

Characteristics of CO₂ plasmolysis

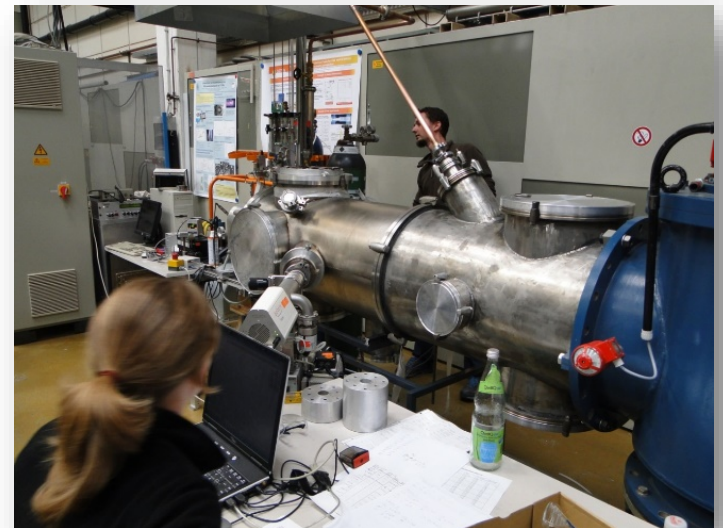
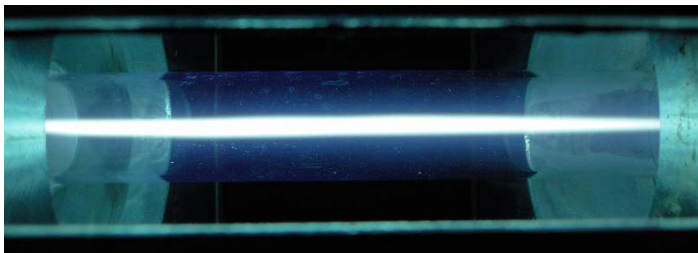
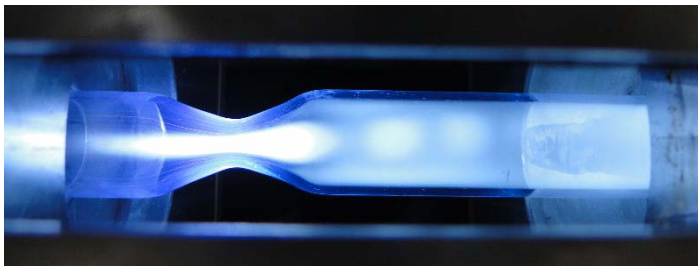
Ease conditions for CO₂ splitting by channelling energy in molecular vibration to break chemical bond, not to heat the gas (non-equilibrium)

- Energy efficiency comparable to Electrolysis (~60% demonstrated)
- High gas flow and power flow density (45W/cm²)
- Fast dynamic response (intermittent power supply)
- No scarce materials employed (Pt catalyst in PEM)



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IGVP: 30 kW @ 915 MHz





Out of equilibrium $T_{\text{vib}} > T_0$ chemistry

Chemical reaction scheme



followed by reuse energetic **O** radical



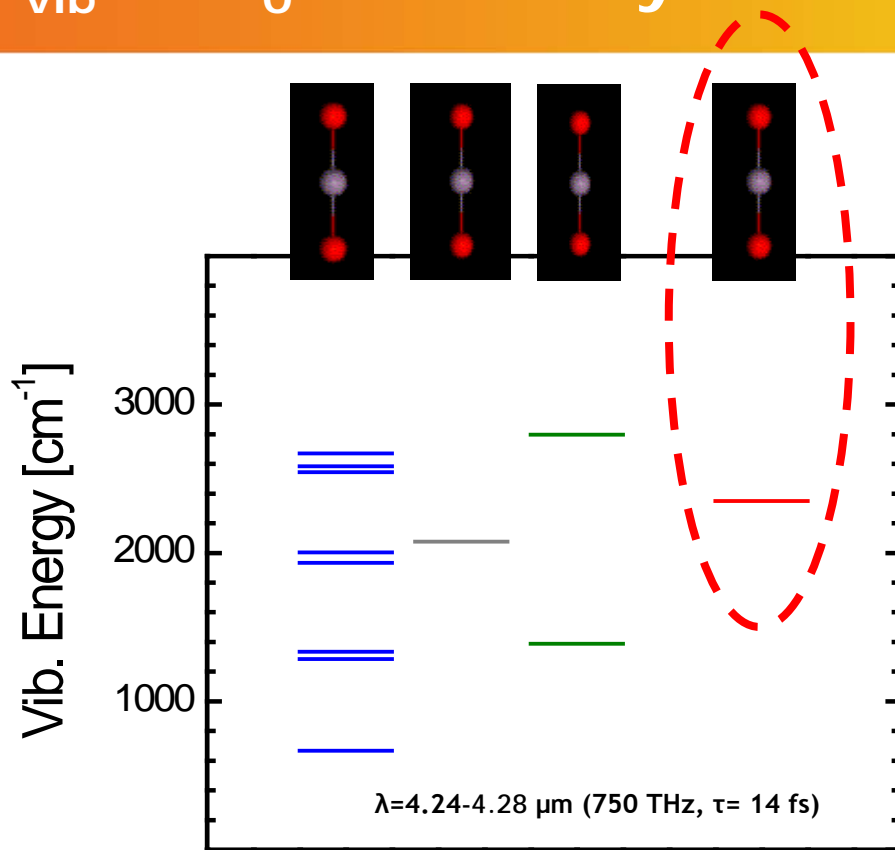
Net



Efficiency to be increased by

Concentration of electron energy on vibrational excitation of CO_2 in asymmetric stretch mode

$$\eta = CF/W = \alpha H/E_v$$



Arrhenius/Fridman:

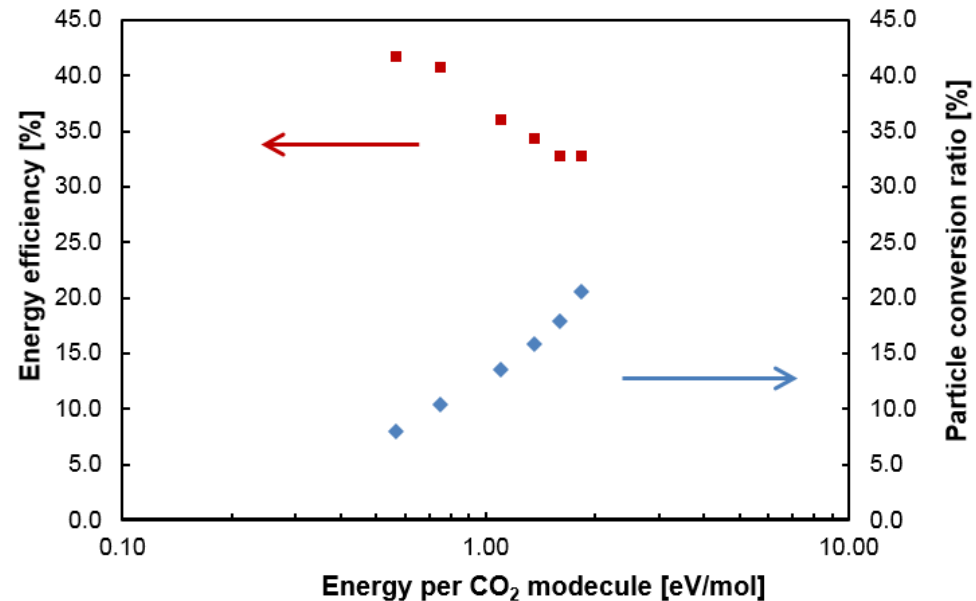
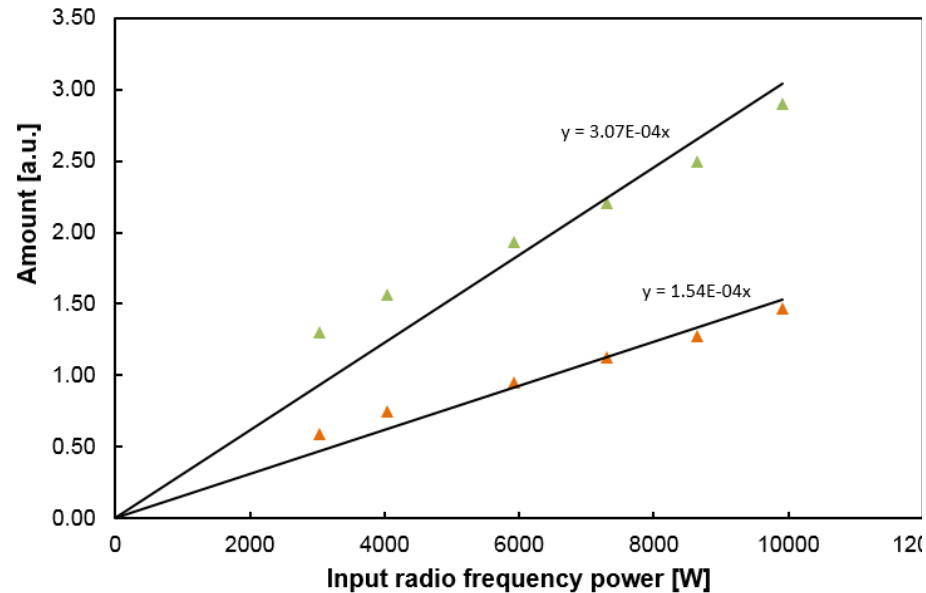
Activation energy reduced by vibration energy

$$k = A \exp(\alpha E_v - E_a) / kT$$



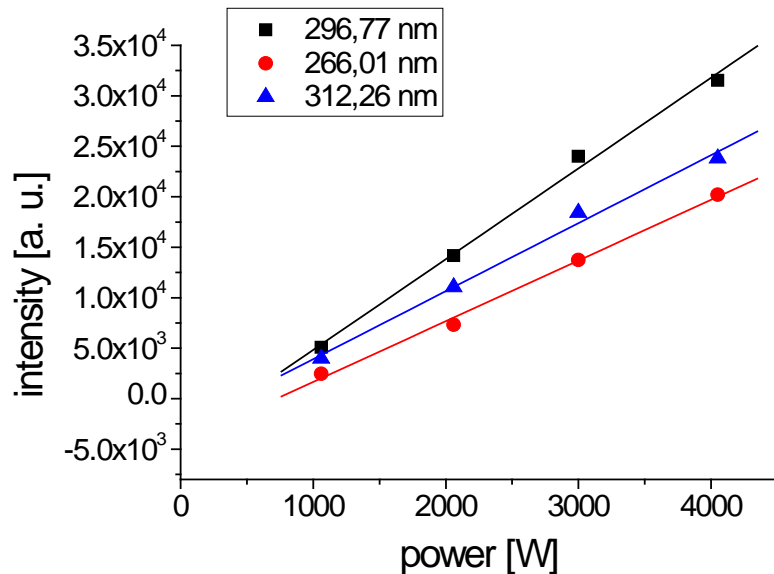
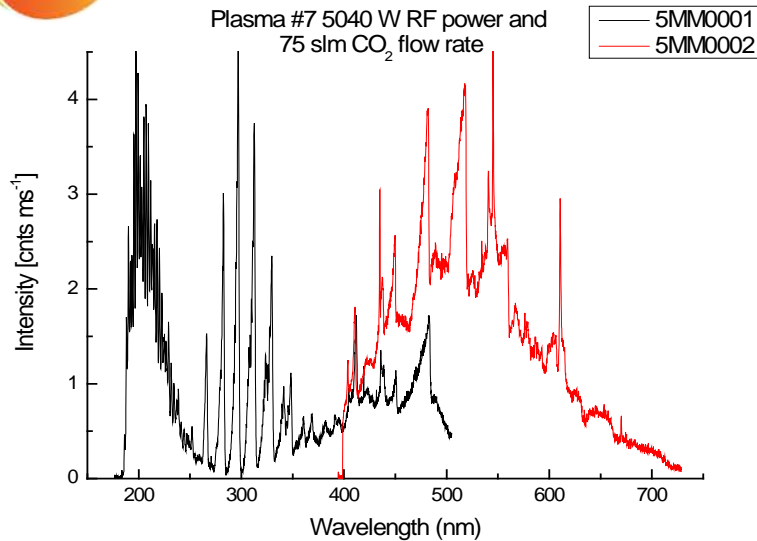
Experimental Results: Mass Spectroscopy

- CO and O₂ production as function **RF Power**

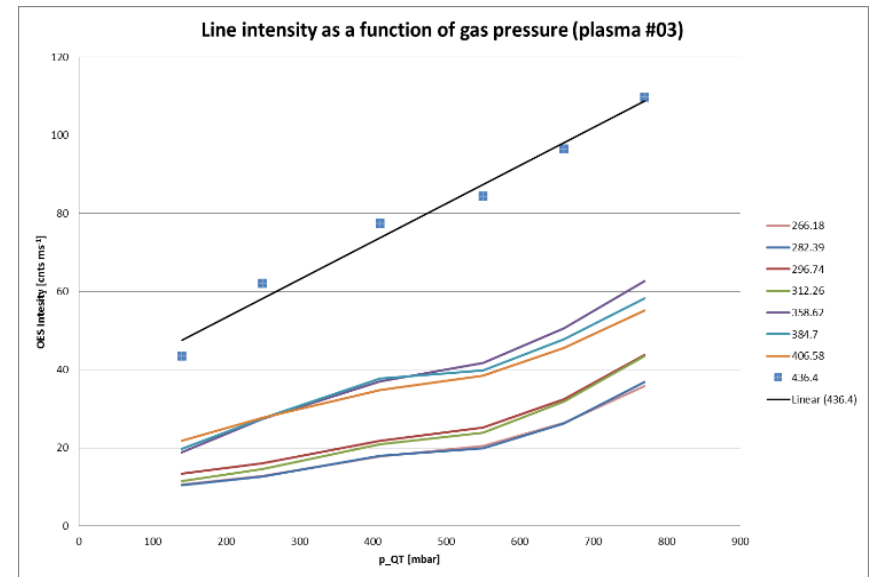




Optical Spectroscopy



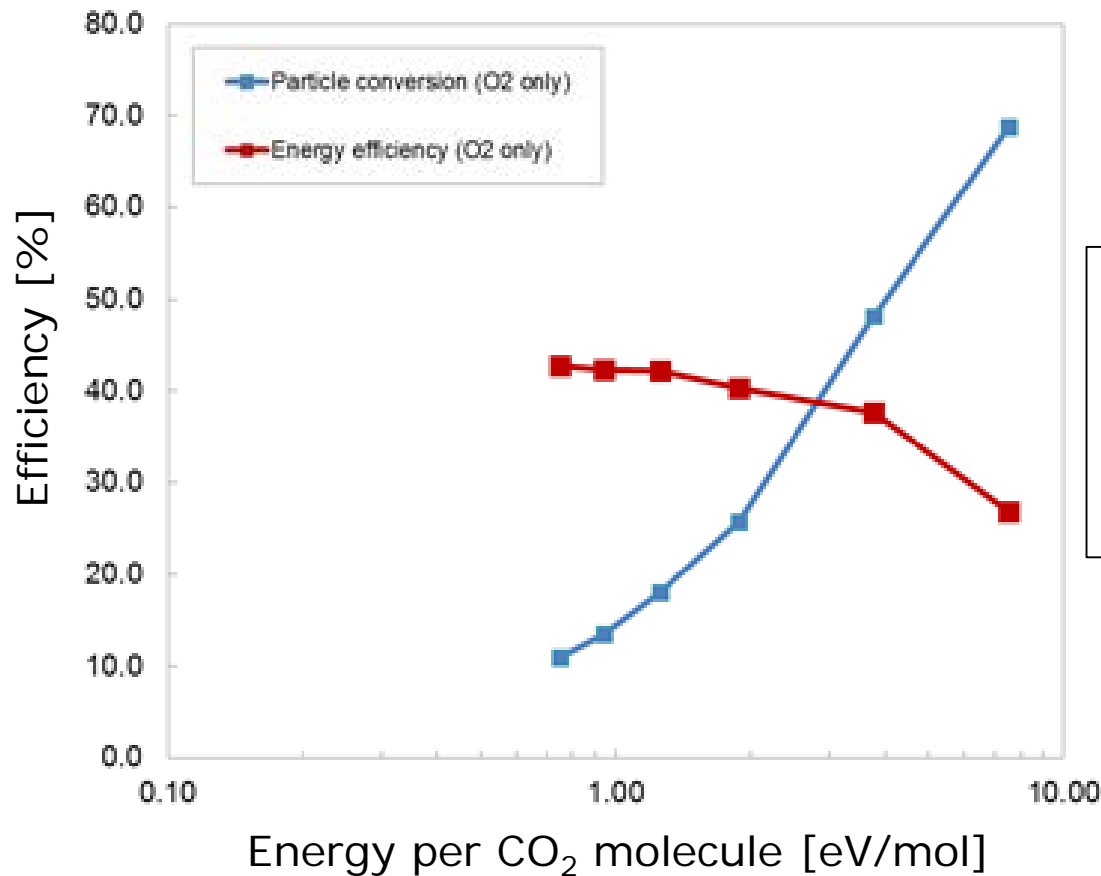
- CO third positives, fourth positives, Angstrom and Triplet identified.
- CO line intensity increases linear with RF power and gas pressure





Experimental Results

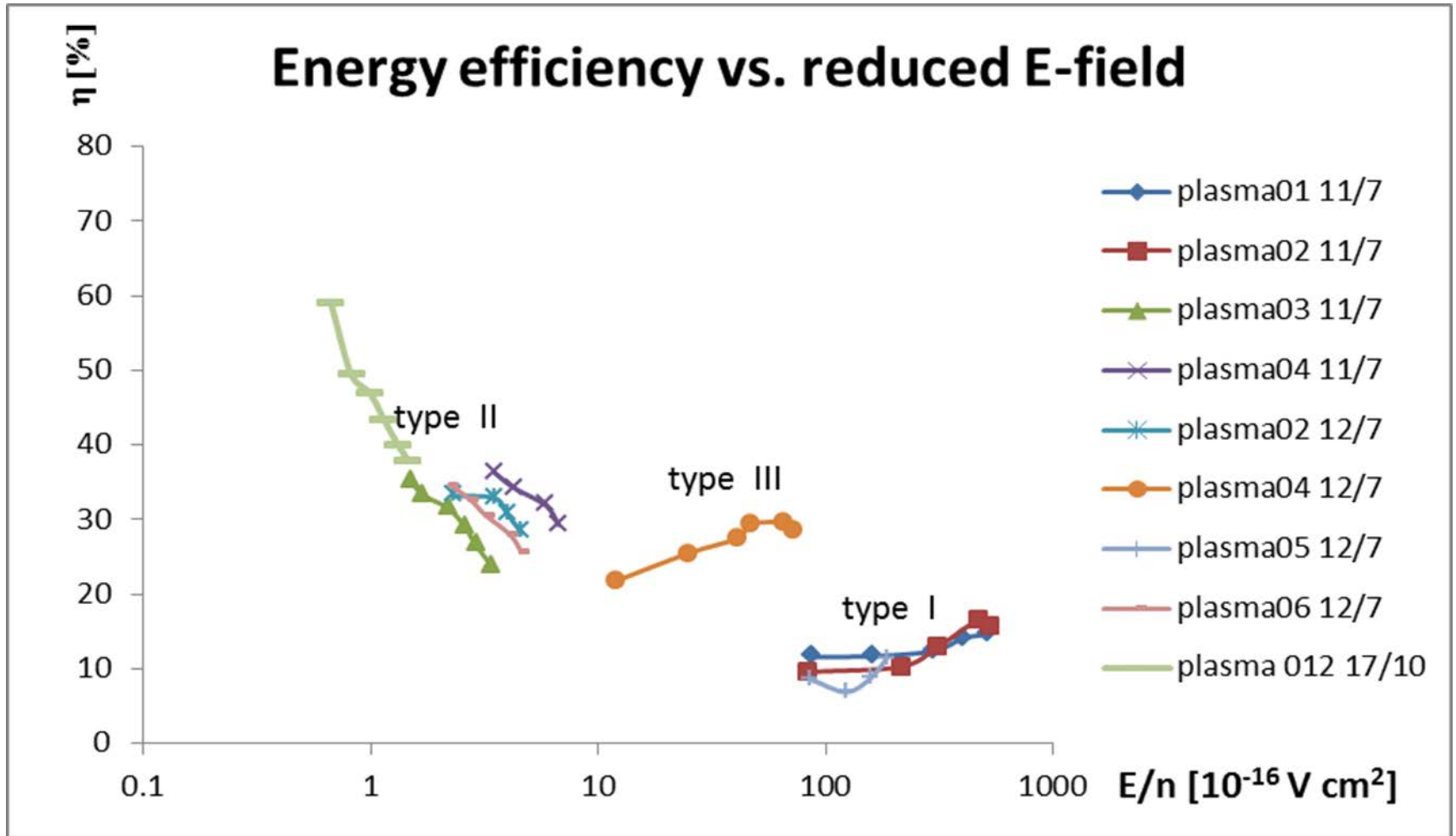
- CO production as function **Gas flow**



10 kW RF absorbed
75 slm CO₂, conversion 10% CO
(non optimised for safety risk)
Pressure 500 mbar,
Energy Efficiency 30%

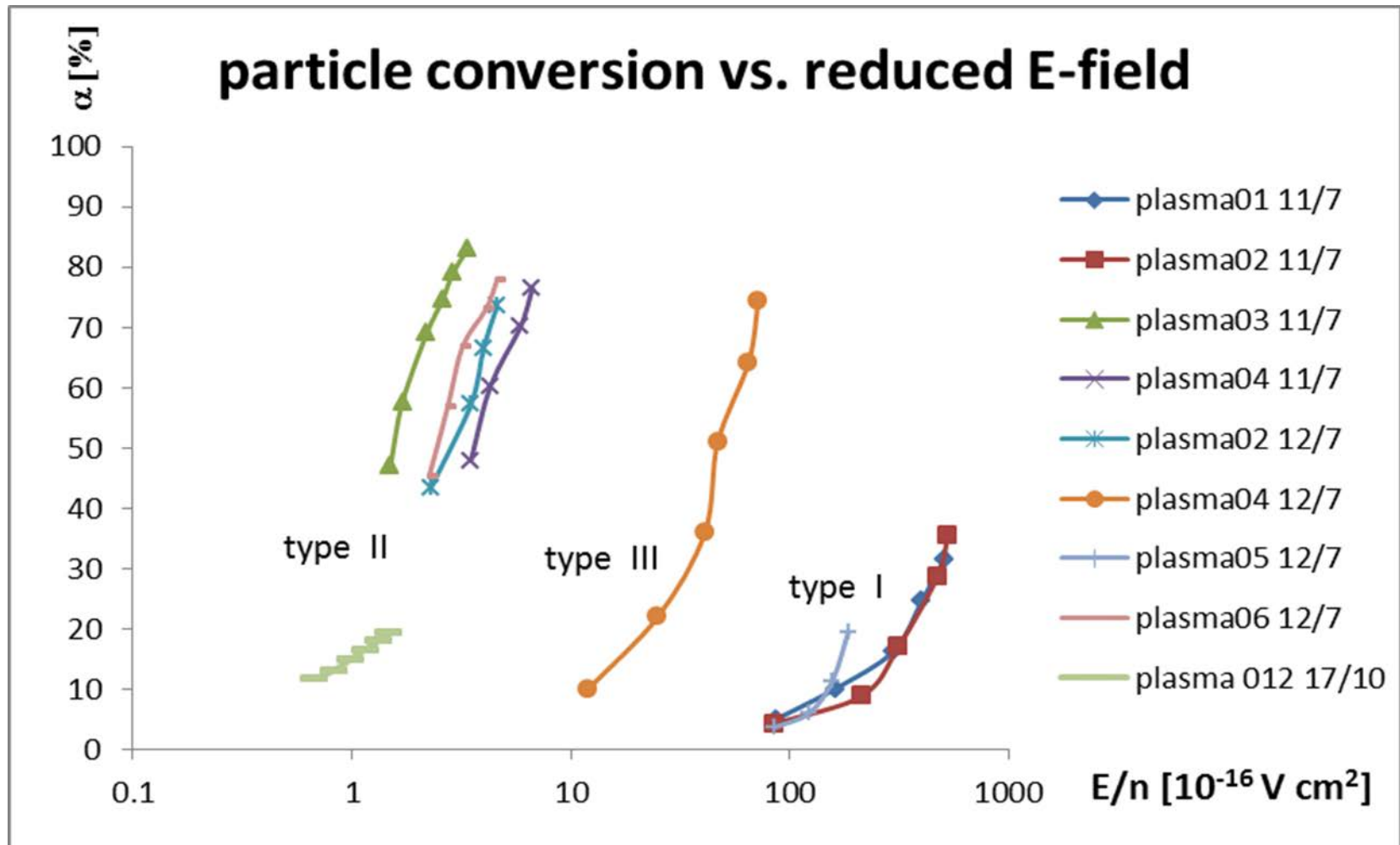


Experimental Results



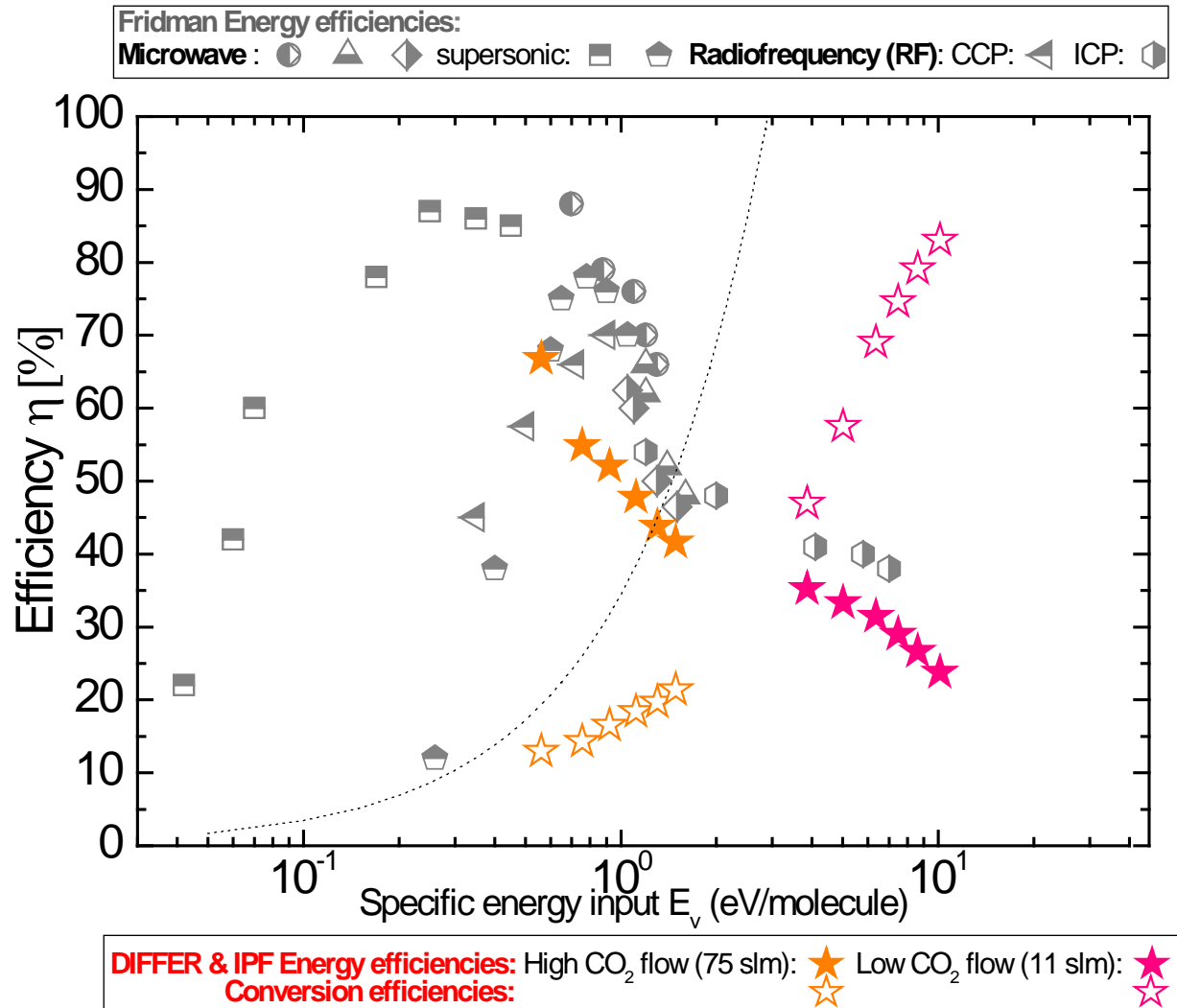


Experimental Results





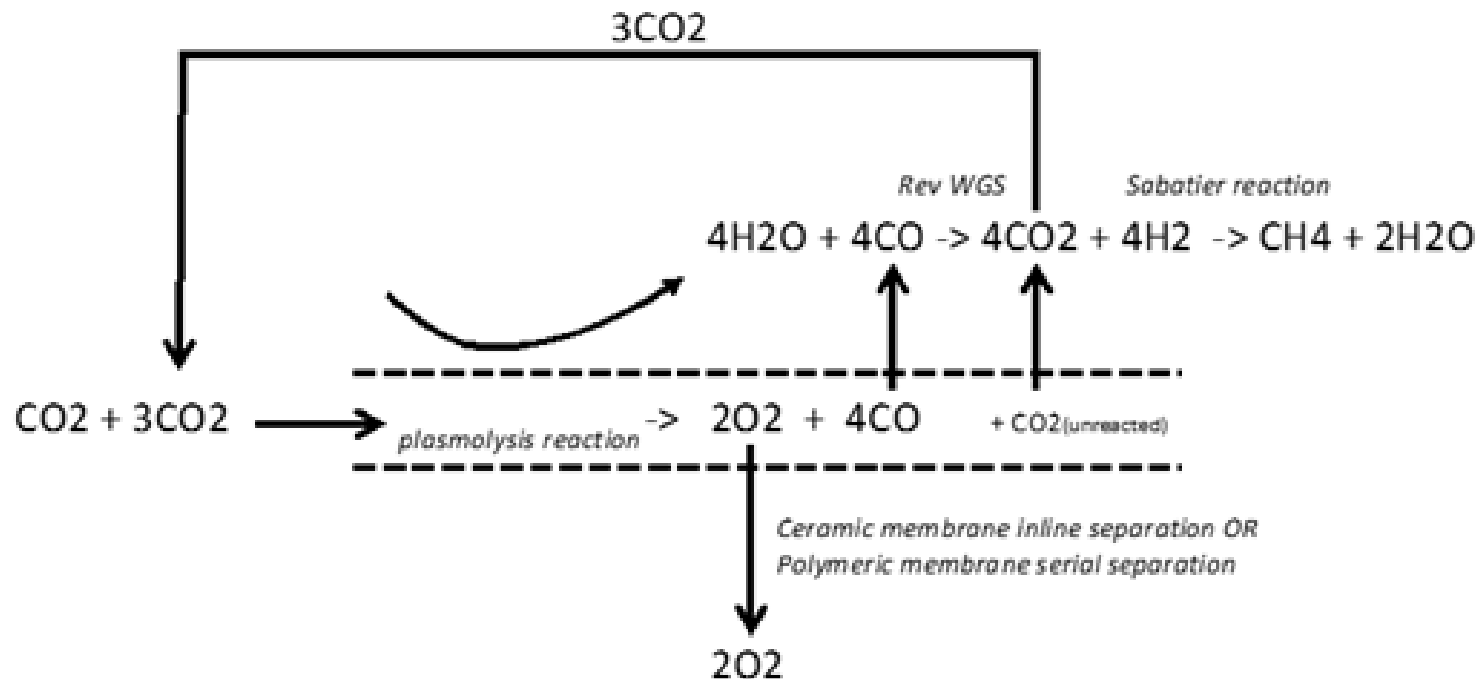
Energy efficiency of CO₂ plasma conversion





O₂ separation from CO (similar sized)

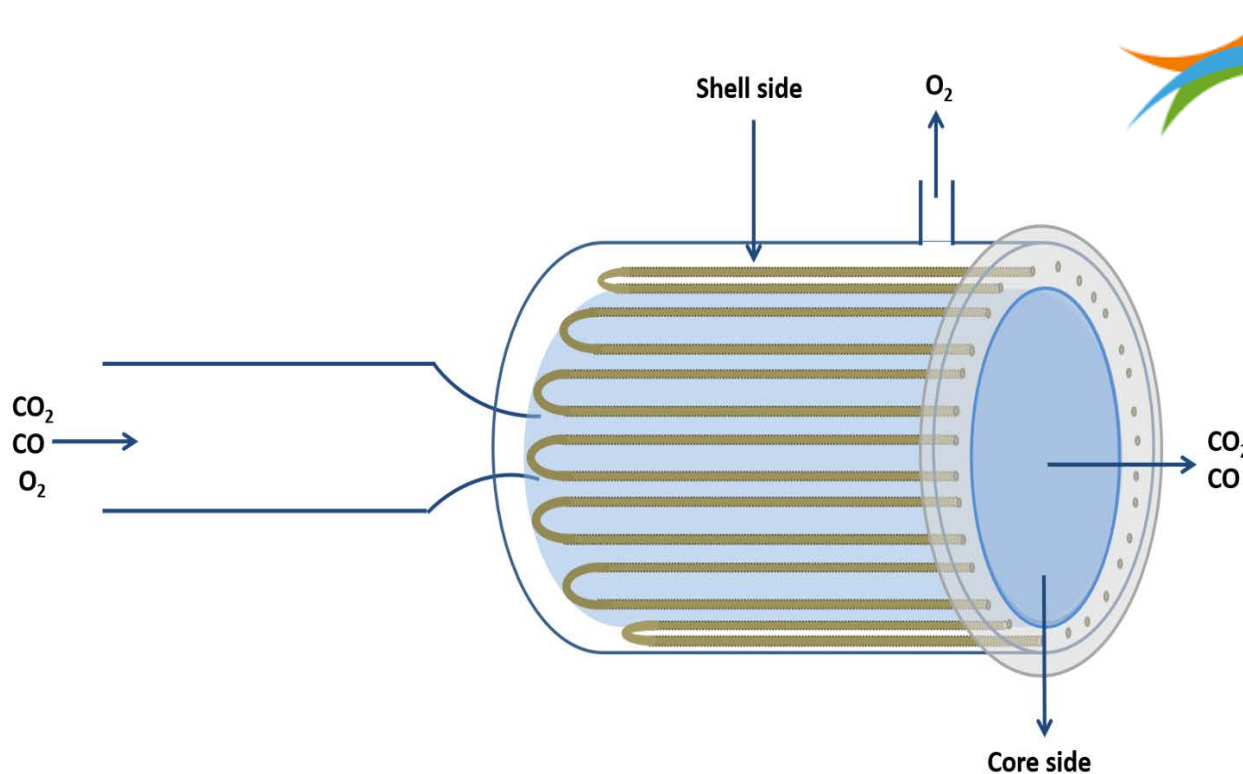
- MIEC mixed ion electron conductive membrane (pressure driven) BSCF (Ba_{0.5} Sr_{0.5} Co_{0.8} Fe_{0.2} O_{3-d}) has been shown to produce an O₂ flux of 60-80 ml/cm²per min.
- Electro chemical Oxygen pump (Voltage driven) YSZ (Yttrium stabilized Zirconia).





Separation of CO, O₂, CO₂ mixture

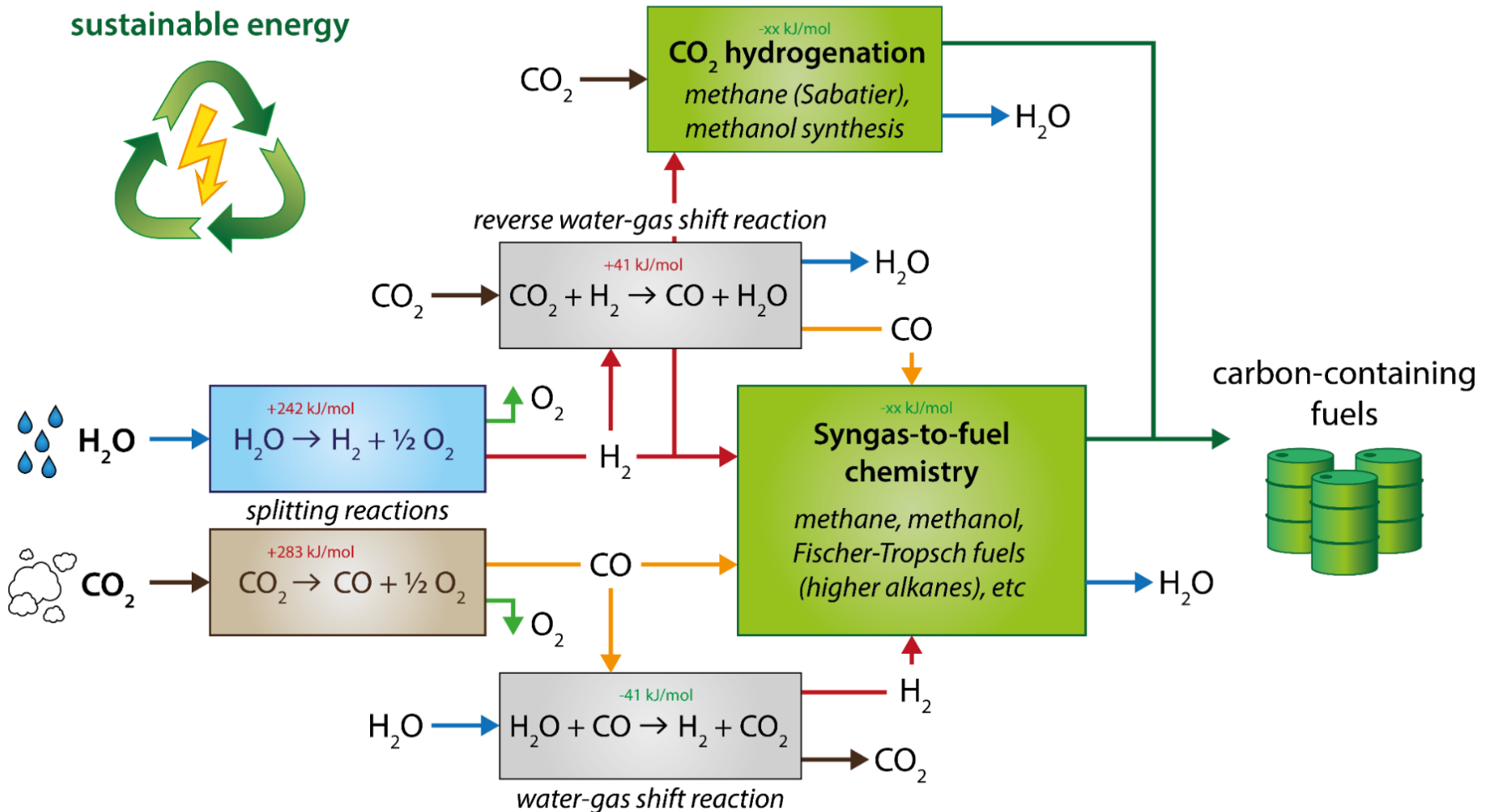
YSZ Oxygen selective membrane to separate O₂ from CO, CO₂ mixture
Hairpin shaped membranes fitted into SS assembly





From H₂O and CO₂ to sustainable hydrocarbons

sustainable energy



reaction enthalpies calculated for gaseous products at standard conditions



10 year Kopernikus Program Power-to-X

Objectives high CO₂ reduction with max CO₂ valorisation, decentralisation, upscaling, public acceptance and export

Yr 1 to 3

- 6 Research Clusters; FZ Jülich, RWTH Aachen, Dechema together with Associated Partners (KIT Energy Lab 2.0)
- Electrolysis, Catalysis, Material and Process design to produce Hydrogen and Syngas
- P2X Road mapping together with Socio-Economic groups and SME

Yr 4 to 7

- 4 Technology Clusters for validation, integration and Pilot projects

Yr 8 to 10

- 3 System Demonstration and Application (Industry lead)



Conclusions

- P2X provides vast seasonal energy storage capacity and provides flexibility in the energy supply from Renewables
- P2X-CCU enables a CO₂ neutral fuel cycle based on hydrocarbons and existing infrastructure
- Technical challenge: innovation in CO₂ splitting and CO-O₂ separation
- Economic challenge: cost reduction, government regulation, business case expected to emerge around 2030, cost of CO₂ to reach € 200/tonne
- In 10 yrs time, cost Wind and Solar tumble, whilst cost of decommissioning, waste disposal and climate change factored into cost of nuclear and fossil power.
- System approach is essential in future energy system, incl. waste disposal, storage and customer acceptance