

Photoelectrochemical CO₂ reduction as a negative emission technology

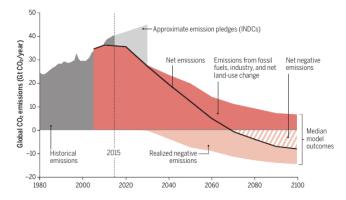
Matthias M. May¹, Kira Rehfeld²

1: Helmholtz-Zentrum Berlin, Institute for Solar Fuels. 2: Universität Heidelberg, Institute of Environmental Physics

> DPG Spring Meeting Rostock 13.03.2019

Negative emissions



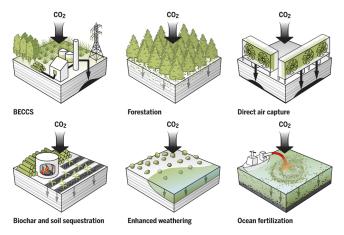


- Anthropogenic emission rates are reduced too slowly
- \rightarrow Almost all climate models assume negative emissions, where energy is invested to sequester atmospheric CO_2, starting from 2030
 - Type of technology and costs still very speculative

[1] Anderson and Peters, Science 354 (2016). [2] Hansen et al., Earth Syst. Dyn. 8 (2017).

Technologies





- Most considered technologies are based on natural photosynthesis
- Sequestration of CO₂ itself relies on (safe) mineral trapping [2]

[1] J. Rosen, Science 359 (2018). [2] Smith et al., Nat. Clim. Change 6 (2016).



Scalable!

- Long-term storage feasible
- Energetic efficiency ca. 2-3% [1]
- \rightarrow Large areas:
- 10 Mio. km² for dedicated crops



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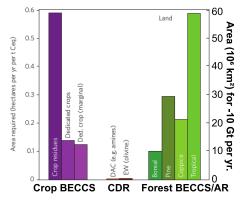
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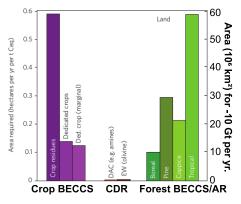


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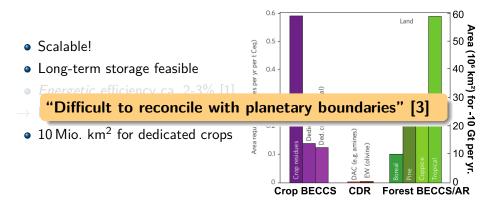


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- (Photo)electrochemical CO₂ reduction
- PV-coupled to dark electrolysis or
- Integrated systems
- ightarrow Challenges of PV & electrocatalysis
- For hydrogen, with 19% STH about 10x more efficient than its natural counterpart [1]



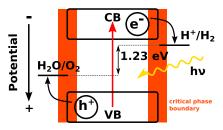


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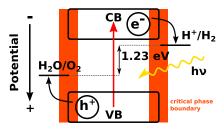


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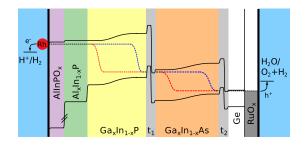


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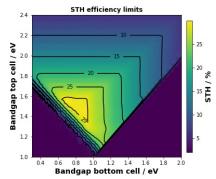


- Multi-junction absorbers required to produce > 1.6 V photovoltage
- Suitable bandgap combinations, efficient catalysis
- Model using detailed balance, $\eta(j)$ from catalysis [2]
- For CO₂ reduction, STF efficiencies are a function of ΔG

[1] May et al., Nat. Comm. 6 (2015). [2] May, et al., Chapter 12 in Integrated Solar Fuel Generators, RSC (2018).

High efficiency



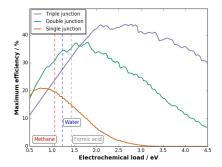


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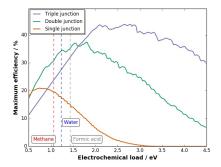


- Solar-to-Fuel efficiencies: $STF = \frac{\eta_F j_e \Delta G}{P \times e}$
- \rightarrow Not suitable for neg. emissions
- Solar-to-carbon (STC) [2]: $STC = \frac{\eta_F \eta_{eje}}{j_{ph}}$
- Large impact of η_e on STC



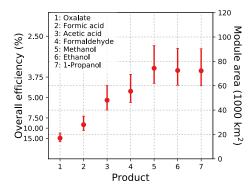


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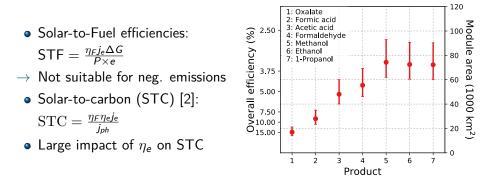




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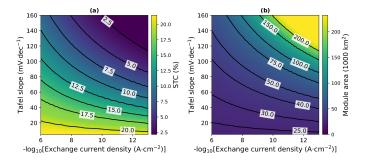




 $e\ {\rm per}\ {\rm carbon}:\ 1\ {\rm for}\ {\rm oxalate};\ 2\ {\rm for}\ {\rm formic}\ {\rm acid};\ 4\ {\rm for}\ {\rm acetic}\ {\rm acid}\ \&\ {\rm formaldehyde};\ 6\ {\rm for}\ {\rm methanol},\ {\rm thanol},\ 1\ {\rm propanol}$

Required area





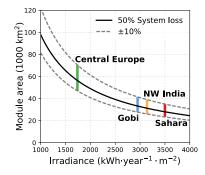
• Catalyst performance crucial

- Desert areas interesting due to high irradiance
- Water consumption (formic acid): ca. 5 km³ as opposed to > 2000 km³ for biomass [2]

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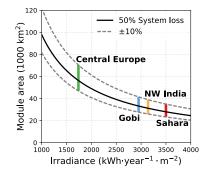


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Storage

- Could be alleviated by chemical post-processing, e.g. oxalate to organic minerals [1]
- Electrochemical production of solid carbon also feasible [2]

Catalysts



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- Natural photosynthesis offers ease of storage, but requires large areas
- Artificial photosynthesis reduces land and water footprint, but will probably be expensive → Solar-To-Carbon efficiency as benchmark
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Thanks for your attention!

Financial support:

